

University of Groningen

Facile Retro Dieckmann Reactions of 3-Oxo-Carbapenam Esters

Vries, Johannes G. de; Hauser, Gerhard; Sigmund, Gerhard

Published in:
 HETEROCYCLES

DOI:
[10.3987/R-1985-05-1081](https://doi.org/10.3987/R-1985-05-1081)

IMPORTANT NOTE: You are advised to consult the publisher's version (publisher's PDF) if you wish to cite from it. Please check the document version below.

Document Version
 Publisher's PDF, also known as Version of record

Publication date:
 1985

[Link to publication in University of Groningen/UMCG research database](#)

Citation for published version (APA):

Vries, J. G. D., Hauser, G., & Sigmund, G. (1985). Facile Retro Dieckmann Reactions of 3-Oxo-Carbapenam Esters. *HETEROCYCLES*, 23(5). <https://doi.org/10.3987/R-1985-05-1081>

Copyright

Other than for strictly personal use, it is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), unless the work is under an open content license (like Creative Commons).

The publication may also be distributed here under the terms of Article 25fa of the Dutch Copyright Act, indicated by the "Taverne" license. More information can be found on the University of Groningen website: <https://www.rug.nl/library/open-access/self-archiving-pure/taverne-amendment>.

Take-down policy

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Downloaded from the University of Groningen/UMCG research database (Pure): <http://www.rug.nl/research/portal>. For technical reasons the number of authors shown on this cover page is limited to 10 maximum.

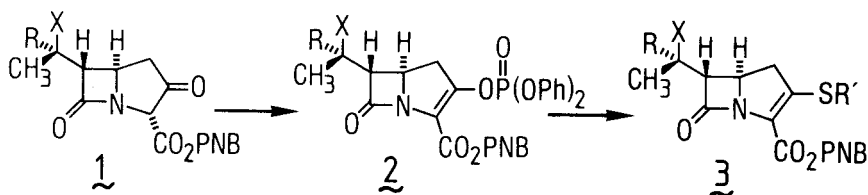
FACILE RETRO DIECKMANN REACTIONS OF 3-OXO-CARBAPENAM ESTERS

Johannes G. de Vries* ¹, Gerhard Hauser, and Gerhard Sigmund

SANDOZ Forschungsinstitut, Brunnerstrasse 59, A-1235 Vienna, Austria

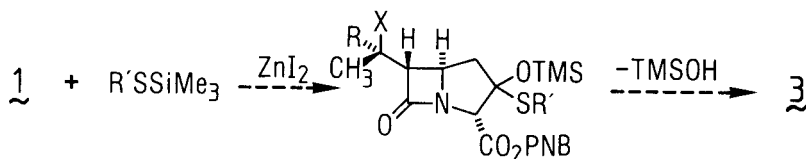
Abstract - Bicyclic ketones **1b** and **1c** reacted with nucleophiles to give azetidinones **4**. Azetidinone **4a** was deprotected to give **5**, which was antibacterially inactive.

The bicyclic ketone **1a** was developed by Merck chemists as a central intermediate for the derivation of carbapenem antibiotics². After conversion to the appropriate enol derivative **2**, preferably the diphenyl phosphoenolate, reaction with a suitable mercaptan under base catalysis gives the 3-thio substituted carbapenem ester **3** (Scheme 1). It occurred to us that a simpler way of achieving this conversion might be by reacting **1** with silylated mercaptans under mild Lewis acid catalysis³ as depicted in Scheme 2.



- a. R=H, X=OP (P=protective group)
 b. R=H, X=F
 c. R=CH₃, X=F

Scheme 1



Scheme 2

In the event, reaction of 1c⁴ with methyl trimethylsilyl sulfide catalyzed by dried ZnI₂ under a variety of conditions was unsuccessful. On the other hand, catalysis by KF/18-crown-6 resulted in a fast and clean reaction at room temperature. However, the product of this reaction was not the hoped for carbapenem ester 3 as was immediately clear from its UV-spectrum lacking the characteristic maximum around 290-300 nm⁵. The β-lactam absorption in the IR at 1754 cm⁻¹ rather suggested a monocyclic β-lactam; an additional absorption at 1680 cm⁻¹ pointed towards a thiol ester. The presence of two diastereotropic protons with a relatively high J_{AB} of 18 Hz in the proton NMR, together with a molecular ion of 412 in the mass spectrum unequivocally establishes structure 4a for this product.

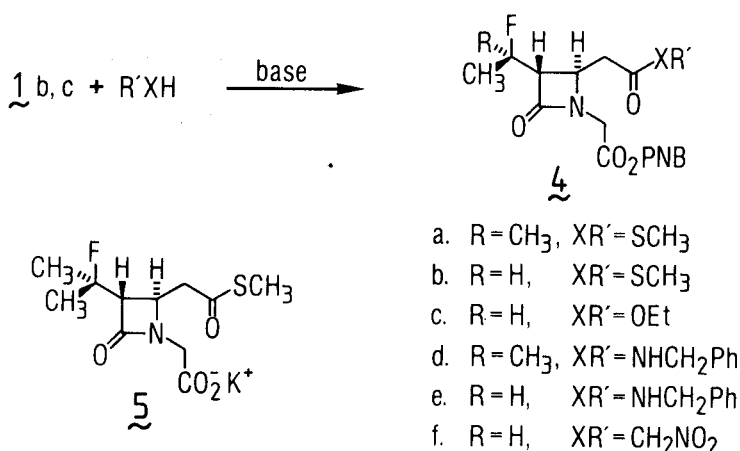
It appears that the more electrophilic site in this highly strained bicyclic system is the ketone moiety, rather than the β-lactam carbonyl⁶. To establish the generality of this observation we reacted ketones 1a and b with some assorted nucleophiles (Scheme 3, see Table for details).

Table. Reactions of bicyclic ketones 1b and 1c with nucleophiles

Entry	Starting material + 4	Nucleophile	Conditions	Product mp ^o C	Yield ⁺⁺
1.	<u>1c</u>	MeSSiMe ₃ , cat. KF, 18-c-6	CH ₂ Cl ₂ , RT 3.5 h	<u>4a</u> , oil	64 %
2.	<u>1b</u>	MeSSiMe ₃ cat. KF, 18-c-6	CH ₂ Cl ₂ , RT 3.5 h	<u>4b</u> , oil	77 %
3.	<u>1b</u>	EtOH, cat. pyridine	EtOH/CH ₂ Cl ₂ 1:1, RT, 16 h	<u>4c</u> , oil	83 %
4.	<u>1c</u>	PhCH ₂ NH ₂	CH ₂ Cl ₂ , RT, 1 h	<u>4d</u> , oil	55 %
5.	<u>1b</u>	PhCH ₂ NH ₂	CH ₂ Cl ₂ , RT, 0.5 h	<u>4e</u> , 75-78 ^o	87 %
6.	<u>1b</u>	CH ₃ NO ₂ , KO ^t Bu	CH ₃ NO ₂ , RT 16 h	<u>4f</u> , 145-147 ^o	42 %

⁺ All compounds are racemic mixtures.

⁺⁺ Yields of chromatographed (silica) products.



Scheme 3

Reaction with benzylamine was rapid and led to the azetidinone amides 4d and 4e⁷. Reaction of 1b with ethanol catalyzed by a small amount of pyridine proceeded overnight and gave the azetidinone ester 4c. A carbon nucleophile was found in the potassium salt of nitromethane, which reacted overnight with 1b to give nitroketone azetidinone 4f in moderate yield.

Not much is known about the biological activity of monocyclic β -lactams having an acetic acid group on nitrogen as recognition site; closest analogues are the nocardicins and they are only moderately active antibacterials⁸. A second detrimental factor is the carbapenem type side chain in the 3-position: N-sulfonated monobactams with such side chains were shown to be devoid of antibacterial activity⁹. Nonetheless, we subjected azetidinone 4a to catalytic hydrogenation (H₂, Pd/C, EtOAc-phosphate buffer pH 7 1:1, 3.5 h) to obtain potassium salt 5 in 36 % yield after RP-18 chromatography (H₂O) and lyophilization of the relevant fractions. As expected the compound did not show any appreciable antibacterial activity (MIC >50 $\mu\text{g/ml}$ in serial dilution test) against a range of gram-positive and gram-negative bacteria, and was a poor β -lactamase inhibitor.

In contrast with the above we found that stabilized phosphoranes react with bicyclic ketone 1 to give C-3 carbon substituted carbapenem esters as a mixture of isomers (endo- and exocyclic double bond at C-3)¹⁰.

ACKNOWLEDGEMENT

We thank Dr. G.Schulz for interpreting NMR spectra and Dr. J.Hildebrandt for *in vitro* testing.

REFERENCES

1. Present address: Sandoz Institute for Medical Research, c/o University College, Gower street, London WC 1E 6BT.
2. a) R.W. Ratcliffe, T.N. Salzmann, and B.G. Christensen, *Tetrahedron Lett.*, 1980, 31.
b) D.G. Melillo, I. Shinkai, T. Liu, K. Ryan, and M. Slettinger, *Tetrahedron Lett.*, 1980, 2783.
3. a) E. Colvin, "Silicon in Organic Synthesis", Butterworths, London, 1981.
b) D.J. Ager, *Chem. Soc. Rev.*, 1982, 11, 493.
4. a) C.P. Mak, K. Wagner, C. Mayerl, and H. Fliri, *Heterocycles*, 1982, 19, 1399.
b) C.P. Mak and H. Fliri, Belgian Patent 897351-A, 1982.
5. R.W. Ratcliff, G. Albers-Schönberg in "Chemistry and Biology of Beta-Lactam Antibiotics", R.B. Morin and M. Gorman, eds., Vol. 2, p. 227, Academic Press, New York, 1982.
6. The reverse reaction has been reported: M. Hatanaka, Y. Yamamoro, H. Nitta, and T. Ishimaru, *Tetrahedron Lett.*, 1981, 3883. These authors used the Dieckmann condensation to synthesize a 3-oxo-carbapenam ester.
7. Spectral Data (^1H and ^{13}C NMR spectra in CDCl_3 , except **5** (D_2O); IR spectra in CH_2Cl_2): **4a**: ^1H NMR: δ 8.24 and 7.55 (4H, AA'BB', J = 9.0 Hz, Ar), 5.27 (2H, s, CH_2Ar), 4.23 (1H, ddd, J = 7.7, 4.8, 2.5 Hz, H-4), 4.24 and 4.03 (2H, AB-q, J = 18.0 Hz, NCH_2), 3.09 (1H, dd, J = 20.0, 2.5 Hz, H-3), 3.04 and 2.99 (2H, ABX, J_{AB} = 16.2 Hz, J = 7.7, 4.8 Hz, CH_2COS), 2.28 (3H, s, SCH_3), 1.53 (3H, d, J = 21.5 Hz, CH_3), 1.43 (3H, d, J = 21.5 Hz, CH_3). ^{13}C NMR: δ 11.7 (SCH_3), 23.89 (CH_3 , J_{CF} = 24.3 Hz), 26.60 (CH_3 , J_{CF} = 24.1 Hz), 42.43 (CH_2COS), 47.24 (NCH_2), 51.81 (C-4), 64.51 (C-3, J_{CF} = 24.3 Hz), 65.56 (OCH_2), 92.71 (CF, J_{CF} = 169.3 Hz), 123.85, 128.85, 142.4, 144.9 (PNB), 165.67 (C-2, J_{CF} = 9.0 Hz), 167.96 (CO_2), 197.18 (COS). IR: 1754, 1680 cm^{-1} . UV (CH_3OH) λ_{max} : 263 (ϵ 700), 238 (ϵ 2900), 224 (ϵ 2400) nm. **4c**: ^1H NMR: δ 8.24 and 7.54 (4H, AA'BB', J = 9.0 Hz, Ar), 5.26 (2H, s, CH_2Ar), 4.97 (1H, ddq, J = 48.5, 7.2, 5.9 Hz, CF), 4.24 and 4.12 (2H, AB-q, J = 18.0 Hz, NCH_2), 4.18 (1H, ddd, J = 8.5, 4.7, 2.4 Hz, H-4), 4.10 and 4.09 (2H, 2 q, J = 7.2 Hz, CH_2CH_3), 3.08 (1H, ddd, J = 18.0, 7.3, 2.4 Hz, H-3), 2.82 and 2.76 (2H, ABX, J_{AB} = 17.2 Hz, J = 8.5, 4.7 Hz, CCH_2CO_2), 1.48 (3H, dd, J = 24.3, 5.9 Hz, CFCH_3), 1.24 (3H, t, J = 7.2 Hz, CH_2CH_3). IR: 1767, 1734 cm^{-1} . **4e**: ^1H NMR: δ 8.20 and 7.50 (4H, AA'BB', J = 9.0 Hz, ArNO_2), 7.22-7.37 (5H, m, Ph), 5.93 (1H, br t, J = 5.5 Hz, NH), 5.18 and 5.20 (2H, AB-q, J = 13.4 Hz, CO_2CH_2), 4.96 (1H, ddq, J = 48.1, 7.5, 6.5 Hz, CF), 4.39 (2H, d, J = 5.5 Hz, NHCH_2), 4.24 (1H, ddd, J = 8.3, 4.3, 2.3 Hz, H-4), 4.21 (2H, s, NCH_2CO_2), 3.06 (1H, ddd, J = 17.3, 7.5, 2.3 Hz, H-3), 2.76 and 2.69 (2H, ABX, J_{AB} = 15.5 Hz, J = 8.3, 4.3 Hz, CH_2CON), 1.47 (3H, dd, J = 24.4, 6.5 Hz, CH_3). IR: 1765, 1677 cm^{-1} . **4f**: ^1H NMR: δ 8.24 and 7.54 (4H, AA'BB', J = 9.0 Hz, Ar), 5.53 and 5.47 (2H, AB-q, J = 17.5 Hz, CH_2Ar), 5.26 (2H, s, CH_2NO_2), 4.98 (1H, ddq, J = 48.4, 7.0, 6.5 Hz, CF), 4.30 and 4.02 (2H, AB-q, J = 18.1 Hz, NCH_2CO_2), 4.26 (1H, ddd, J = 7.5, 5.5, 2.3 Hz, H-4), 3.12-3.19 (2H, m, CH_2COC), 3.12 (1H, ddd, J = 19.2, 7.0, 2.3 Hz, H-3), 1.46 (3H, dd, J = 23.3, 6.5 Hz, CH_3). IR: 1769, 1751 cm^{-1} . **5**: ^1H NMR: δ 4.23 (1H, ddd, J = 7.2, 5.9, 2.3 Hz, H-4), 3.83 (2H, s, CH_2COS), 3.37 (1H, dd, J = 24.2, 2.3 Hz, H-3), 3.22 and 3.08 (2H, ABX, J_{AB} = 15.6 Hz, J = 7.2, 5.9 Hz, CH_2COS), 1.52 (3H, d, J = 20.6 Hz, CH_3), 1.43 (3H, d, J = 20.6 Hz, CH_3). IR (KBr): 1732, 1669 cm^{-1} . UV (H_2O) λ_{max} : 234 (ϵ 1400), 218 (ϵ 900) nm.
8. T. Kamiya, H. Aoki, and Y. Mine, p. 165 in Ref. 5.
9. C.J. Ashcroft, J. Brennan, C.E. Newall, and S.M. Roberts, *Tetrahedron Lett.*, 1984, 877.
10. J.G. de Vries, G. Hauser, and G. Sigmund, *Tetrahedron Lett.*, 1984, 5989.

Received, 14th January, 1985